# organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

## Amir H. Mahmoudkhani and Ignacio Vargas-Baca\*

Department of Chemistry, McMaster University, Hamilton, Ontario, Canada L8S 4M1

Correspondence e-mail: vargas@chemistry.mcmaster.ca

#### Key indicators

Single-crystal X-ray study T = 173 KMean  $\sigma(C-C) = 0.004 \text{ Å}$  R factor = 0.044 wR factor = 0.114 Data-to-parameter ratio = 16.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# *N*-(*tert*-Butyl)-*S*-(4-methylphenyl)thiohydroxylamine

The title compound,  $(CH_3)_3CN(H)-SC_6H_4CH_3-4$ , is the first example of a simple alkylarylsulfenamide to be structurally characterized. Crystal packing consists of ribbons formed by  $N-H\cdots S$  hydrogen bonds.

Received 20 June 2003 Accepted 25 June 2003 Online 30 June 2003

## Comment

Sulfenamides, RN(H)–SR' (R, R' = alkyl or aryl), are compounds with simple structures yet remarkably rich chemistry. Their photolysis or chemical oxidation results in the formation of free radicals, some of which are persistent for long periods of time while others decay through complicated pathways (Miura, 1997). Deprotonation yields anions that act as ambidentate ligands towards transition (Hankin et al., 1995, 1996a,b; Danopoulos et al., 2000) and main-group metal ions (Mahmoudkhani et al., 2003). There are only two compounds that have been structurally characterized previously, bearing the sulfenamide functional group as part of a more complex molecule (Lee et al., 1995; Gotthardt et al., 1987). However, the structures of the simplest members of the family were not examined until now, partly because of their low melting points. Highly pure (p-tolyl)(tert-butyl)sulfenamide, (I), solidifies just below room temperature, which allowed this study. The compound crystallizes in the monoclinic space group C2/c(No. 15).



The molecular structure is shown in Fig. 1. It contains an N atom as a chiral center, though the material is a racemic mixture. The C–N–S–C center adopts a *gauche* conformation with a torsion angle of  $-113.8 (2)^{\circ}$ . The (N–)H atoms



#### Figure 1

© 2003 International Union of Crystallography Printed in Great Britain – all rights reserved

The molecular structure of (I). Displacement ellipsoids are drawn at the 50% probability level.



Figure 2

Representation of the ribbon structure of (I). The (C-)H atoms have been omitted for clarity.

are engaged in intermolecular hydrogen bonds with the S atoms, forming a ribbon, self-assembled along the b axis (Fig. 2).

## **Experimental**

The synthesis of (I) was described previously (Mahmoudkhani et al., 2003). Crystals were grown from the melt upon standing at room temperature overnight.

### Crystal data

$D_x = 1.128 \text{ Mg m}^{-3}$
Mo $K\alpha$ radiation
Cell parameters from 833
reflections
$\theta = 1-24^{\circ}$
$\mu = 0.24 \text{ mm}^{-1}$
T = 173 (2) K
Needle, colorless
$0.40 \times 0.12 \times 0.02 \text{ mm}$
1359 reflections with $I > 2\sigma(I)$

 $\omega$  scans Absorption correction: multi-scan (SADABS; Sheldrick, 2001)  $T_{\rm min}=0.910,\ T_{\rm max}=0.996$ 8741 measured reflections 2108 independent reflections

### Refinement

Refinement on  $F^2$  $R[F^2 > 2\sigma(F^2)] = 0.044$ wR(F<sup>2</sup>) = 0.114 S = 1.012108 reflections 130 parameters H atoms treated by a mixture of independent and constrained refinement

 $R_{\rm int}=0.066$  $\theta_{\rm max} = 25.4^\circ$  $h = -27 \rightarrow 29$  $k = -7 \rightarrow 7$  $l = -19 \rightarrow 19$ 

 $w = 1/[\sigma^2(F_o^2) + (0.0538P)^2]$ + 0.5048P] where  $P = (F_o^2 + 2F_c^2)/3$  $(\Delta/\sigma)_{\rm max} = 0.008$  $\Delta \rho_{\rm max} = 0.23 \text{ e} \text{ Å}^{-3}$  $\Delta \rho_{\rm min} = -0.26 \text{ e } \text{\AA}^{-3}$ 

Table 1	
i aoic i	

Selected geometric parameters (Å, °).

\$1-N1	1.694 (2)	N1-C7	1.486 (3)
S1-C1	1.776 (3)	N1-H1	0.82 (2)
N1-S1-C1	103.57 (11)	C7-N1-H1	111.4 (16)
C7-N1-S1	118.43 (16)	\$1-N1-H1	110.4 (16)

### Table 2

#### Hydrogen-bonding geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$N1 - H1 \cdots S1^i$	0.82 (2)	2.75 (2)	3.523 (3)	159 (2)
Symmetry code: (i) <sup>1</sup> / <sub>2</sub>	$\frac{1}{2} - x, y - \frac{1}{2}, \frac{1}{2} - z$			

The (N-)H atom was located from a difference Fourier map, whereas the (C-)H atoms were placed 0.98 Å from the parent atom with bond angles constrained to idealized values using the appropriate riding model and refined isotropically;  $U_{iso}$  values are in the range 0.020-0.073 Å<sup>2</sup>.

Data collection: SMART (Bruker, 1997); cell refinement: SMART and SAINT (Bruker, 1997); data reduction: SAINT and SADABS (Sheldrick, 2001); program(s) used to solve structure: SHELXTL (Bruker, 2001); program(s) used to refine structure: SHELXTL; molecular graphics: DIAMOND (Brandenburg, 2001); software used to prepare material for publication: SHELXTL.

## References

- Brandenburg, K. (2001). DIAMOND. Version 2.1e, Crystal Impact GbR, Bonn, Germany,
- Bruker (1997). SMART and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (2001). SHELXTL. Version 6.10. Bruker AXS Inc., Madison, Wisconsin, USA.
- Danopoulos, A. A., Hankin, D. M., Cafferkey, S. M. & Hursthouse, M. B. (2000). J. Chem. Soc. Dalton Trans. pp. 1613-1615.
- Hankin, D. M., Danopoulos, A. A., Wilkinson, G., Sweet. T. K. N. & Hursthouse, M. B. (1995). J. Chem. Soc. Dalton Trans. pp. 1059-1060.

Hankin, D. M., Danopoulos, A. A., Wilkinson, G., Sweet. T. K. N. & Hursthouse, M. B. (1996a). J. Chem. Soc. Dalton Trans. pp. 1309-1321.

Hankin, D. M., Danopoulos, A. A., Wilkinson, G., Sweet. T. K. N. & Hursthouse, M. B. (1996b). J. Chem. Soc. Dalton Trans. pp. 4063-4069.

- Gotthardt, H., Lohr, T. & Brauer, D. J. (1987). Chem. Ber. 120, 747-750. Lee, C. K., Linden, A. & Vasella, A. (1995). Acta Cryst. C51, 1906-1910.
- Mahmoudkhani, A. H., Rauscher, S., Grajales, B. & Vargas-Baca, I. (2003). Inorg. Chem. 42, 3849-3855.
- Miura, Y. (1997). Trends Org. Chem. 6, 187-217.
- Sheldrick, G. M. (2001). SADABS. Version 2.03. University of Göttingen, Germany.